EXPERIMENTAL STUDY

Toxic effect of nano-zinc oxide

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ABSTRACT

INTRODUCTION: Nowadays, a lot of studies were carried out about side effects of nanoparticles especially nano-zinc oxide, but unfortunately, most of these studies have been conducted with high doses.

MATERIAL AND METHODS: In the present study 36 male Wistar rats with approximate age of two months and weighing 300 ± 30 g were selected. After standard phases of preparation, injections were done intraperitoneally in chronic groups with zinc oxide nanoparticles (20 and 40 mg/kg) every other day and for acute groups with 200 and 400 mg/kg in a single dose. After the determined time for every group (21 days for chronic groups and 7 days for acute groups), blood samples were provided from rat hearts and also liver, kidney and testicular tissue were removed for histological studies.

RESULTS: There were significant decreases in lymphocytes and platelet counts and significant increases in white blood cells count, hemoglobin and hematocrit in acute dose compared to control group and no difference in red blood cells count in all groups. A severe to slight damage was observed in liver tissue in all groups and tissue destruction was seen in kidney only in acute dose treated groups but no specific pathological effect in testis was observed (Fig. 8, Ref. 43). Text in PDF www.elis.sk.

KEY WORDS: nano-zinc oxide, acute and chronic doses, blood parameters, oxidative stress, liver tissue.

Introduction

Zinc oxide (ZnO) is a nonorganic substance and white powder that is widely used as an additive in different materials and productions including plastics, ceramics, glass, cement, tires, lubricants, dyes, sunscreens, glues, pigments, food and batteries, etc. (1). Negative and destructive imagination about potential damages of nanotechnology, whether real or illusionary may result in a threat and slow the growth and development of nanotechnology, unless it is published correctly, in particular information about its dangers and how to prevent them (2). One of the most important of nano-oxides is nano-zinc oxide. These nanoparticles can substitute the macromolecule zinc oxide in various industries and give special properties to the final product (3). Zinc oxide nanoparticles have globular shape and recently, have received more attention in animal studies (4). Production and usage of different synthetic substances have influenced the health of many organisms. The nanoparticles enter in to body through various ways such as skin, inhaling and food (5). Donaldson and coworkers in 2004 made an evolution in toxicology by introducing the science of toxicology in nanoparticles (6). Despite the widespread uses of nanoparticles, there is limited information about their toxic effects in environment and in humans. Nanoparticles do not have any special problem in passing the physiological barriers due to their small size, so they are widely distributed in the circulation (7–9).

Oxidative stress in biological systems was introduced first by Sies and colleagues. It is an imbalance in oxidant and antioxidant systems in which there is an increase in the oxidant ones that cause damage (10). The production of ROS can disturb mitochondrial function and may also cause changes in the expression of genes that are involved in inflammation and apoptosis.

The science of nanoparticles toxicology will have an important role in the development of stable and safe nanotechnology. Although little information is available about toxic effects of nanoparticles on environment and humans, according to physical-chemical properties of nanoparticles it is expected that these substances interact with biological components and have a large impact on behavior and features of macromolecules, cells and bodies of organisms (11).

In order to study the toxicity of zinc nanoparticles (and other nanoparticles) on different organs a lot of experiments were performed and most of the results are indicating toxicity of them. The largest problem in these studies is the use of high doses in vivo and in vitro researches, although low doses were used in some studies but the number of them is negligible. In the general case the results of many studies indicate toxicity of zinc oxide nanoparticles and its negative effects on the organisms and target cells (12). These nanoparticles can pass from protective layers of body and enter the circulation and cause severe effects (13, 14). Recently, it has
been reported that many concerns exist about the effects of these substances on human health and environment. Previous studies have shown that some nanoparticles produce reactive oxygen species (ROS) and free radicals and cause toxicity in laboratory environments. An imbalance between antioxidant defense and free radicals production causes a condition that is named oxidative stress and may cause cell damage and it is also very important in the pathogenesis of some diseases. The outcomes of oxidative stress depending on cell type and severity of oxidative stress are one of the cell proliferation, cell damage, cell death and compatibility or the combination of them.

The results of studies show that nano-zinc oxide can have both antioxidant effects and also be an oxidative stress factor. It seems this multiple functions are related to the amount of existing material and nanoparticle concentration. Zinc in low doses has beneficial functions in metalloenzymes, transcription factors, immune regulation and growth and synthesis of cell proteins and plays antioxidant, anti-apoptotic and anti-inflammation roles (15, 16). In high doses, it has toxic effects and causes oxidative stress through increasing end products of lipid peroxidation and decreasing glutathione and protein thiol levels. Cytotoxicity increases with rise in nanoparticles concentration and also exposure time (24). The nanoparticles of zinc oxide can pass from protective layers of body; enter the circulation and lead to severe effects especially predispose individuals to thrombosis in high doses (25). Zinc nanoparticles participate in free radical formation in skin and damage cells and make cause cancer (26). Nanoparticles can easily cross from cell membranes and even blood-brain and blood-testes barriers (26, 27, 28).

Liver is the central organ in metabolism and detoxification. Because of that this organ is the target of different toxins that predispose it to many diseases (32). Sharma and colleagues indicated that zinc oxide nanoparticles could induce oxidative damage in DNA and also cause mitochondrial, genotoxic effects and apoptosis in human liver cells (33). In one study, it has been shown that treatment of mice with nanoscale zinc and microscale zinc components resulted in inflammation in the renal glomeruli but there were destructive effects in nano-oxide form and there was no serious damage in testicular tissue observed.

Results

In the present study, we tried to examine the effect of zinc oxide nanoparticles in both acute and chronic doses. For this purpose, two groups of six experimental groups were treated with 200 and 400 mg/kg/every dose. Two other groups were treated with the same doses but they were divided in 10 doses to increase the exposure of zinc nano-oxide but not to increase the final dose. Treatment was done by usual water and food. Second control group that referred to placebo, was injected by 1 ml of distilled water every other day intraperitoneally for equivalency of shock resulting from the intraperitoneal injection. Other groups, 3rd and 4th were injected 1ml ZnO-NPs in 20 and 40 mg/kg doses, injections were repeated every other day intraperitoneally. This continued until 21st day (injection was repeated 10 times). Therefore final dose (20*10 and 40*10) was 200 and 400 mg/kg. 5th and 6th group were treated by 1 ml of ZnO-NPs in 200 and 400 mg/kg doses (single dose) that were considered acute doses. The rats received the same doses in chronic and acute groups. ZnO-NPs resolved in physiological serum in 20 min by sonication method producing a stable suspension. After the mentioned time course blood samples of control and treated rats were prepared and blood levels of hemoglobin, hematocrit, red and white blood cells, lymphocytes, and platelet counts were estimated.

Also, for histological studies liver, kidney and testis were removed and after washing with normal saline for fixation they were placed in Bouin’s solution. Then tissue passaging, preparation of paraffin blocks and cutting were performed. Staining was done with hematoxylin-eosin and finally the different cells were examined with an optical microscope.

Data were analyzed using analysis of variance (ANOVA). In the case of a significant ANOVA, post hoc analysis was performed using Tukey’s test. Difference level of p < 0.05 was considered significant.

Results

In the present study, we tried to examine the effect of zinc oxide nanoparticles on WBC and hemoglobin concentrations shows that there were a number of differences.

Material and methods

Zinc oxide nanoparticles

Zinc oxide nanoparticles were prepared from Tecnan, Spain (CAS NO: 1314-13-2, Particle Average Size: 20–30 nm, Specific Surface Area: 35–50 m²/g, Purity: 99.983 %). Animals

It was an experimental study carried out on animals. We used adult male Wistar rats weighing 250–300 g with average age of 2 months. The rats were housed in groups under standard light conditions with free access to water and food. Humidity and temperature (22 ± 1 °C) were controlled in ventilated cages in a 12-hour day/night cycle. Experimental animals were randomly divided into 6 groups (6 rats in each group). First control group was performed for comparison of chronic exposure against acute exposure.
significant differences among groups receiving nano-zinc oxide compared to control group with 400 mg dose but there were no significant differences in other groups including groups receiving 200 mg dose and chronic dose (Figs 1 and 2).

There were significant differences in lymphocytes, hematocrit and platelets in groups receiving nano-zinc oxide (200 and 400 mg) compared to control group but no significant change was observed in chronic administrations (Figs 3, 4 and 5). The RBC levels variation between experimental groups and the control group were not statistically significant (data not showed).

The findings in four experimental groups (2 chronic and 2 acute groups) showed that zinc nano-oxide has toxic effect on liver.

Discussion

In this study, hemoglobin increased with the 400 mg/kg dose and also hematocrit level elevated in the 200 and 400 mg/kg dose groups. Whereas these parameters are suggesting an increase in blood concentration, they indicate that nano-zinc oxide has a toxic effect on blood viscosity. It seems that nano-zinc oxide can cause inflammation in inner vessels in acute conditions. One of the most important reasons of increasing blood viscosity is a vascular leak that occurs in inflammatory diseases, infectious diseases and burns. Previous studies also have showed blood leakage from vessel walls with high doses of nano-zinc oxide (35, 36). The reason of this can be an increase in expression of VEGF gene that takes place in inflammatory tissues, platelets and immune cells (37–39). Platelets are involved in various aspects of homeostasis. Platelets bind to collagen fibers located beneath the endothelium and prevent bleeding. Reduction in platelet numbers in peripheral circulation results in blood leakage to tissues. Maybe, the effects of oxidative stress on blood platelets and departure of vessels are the other reasons for an increase in hematocrit. Also, white blood cells increased in acute groups. The oxidative stress caused by acute injection of nano-zinc oxide leaded to inflammation in vessels which in turn resulted in WBC accumulation and the release of chemical mediators and further damage to the arteries and spread of inflammation. White blood cells can harm the endothelium by means of releasing cytokines (40, 41).

In liver histological examination the effects were severe in acute doses than chronic doses. In 200 and 400 mg/kg doses focal hyperplasia of Kupffer cells, mild prostatitis and small focal collections of lymphocytes were observed. There were a few small granulo-
lomas in liver parenchyma in 20 and 40 mg/kg doses groups. These findings indicate that liver has been exposed to chronic inflammation that it is consistent to some extent with Cho and coworkers study about nano-zinc oxide effect and granulomatous in liver (42). Kupffer cells have critical role in deleting nanomaterials, so liver tissue is more vulnerable than other organs (43). Considering the measure of liver damage in acute dose to chronic dose in addition to nano-zinc oxide dose, can point to liver tissue tolerance and rapid elimination from blood by other systems such as lymph nodes and spleen-reticular-endothelial system in chronic groups compared to acute groups.

We observed some renal tubular cysts and slight swelling in renal glomerulus with acute doses but there was no clear alteration in renal glomerus with chronic doses. These differences may relate to acute dose of nano-zinc oxide like liver. On the other hand, based on previous studies reticular-endothelial systems of liver and spleen and lymph nodes have essential role in elimination of nanomaterials from circulation. So it is conceivable that there is less damage in kidney than liver that is consistent with our findings. As inflammation can lead to oxidative stress and vice versa, so it can be concluded that oxidative stress caused by nano-zinc oxide is the main factor in inflammatory effects of this substance on liver and kidney.

Finally, no damage was seen in testicular tissue that is probably due to insufficient concentration of nano-zinc oxide for effecting testis. This finding is consistent with Bing Wang’s study that has shown that certain concentrations of nano-zinc oxide have toxic effects on the kidney, but same concentration did not cause visible effects on testicular tissue (34).

**Conclusion**

Nano-zinc oxide has different effects on one organ in different concentrations and different time exposure. Also, all organs are not destroyed at the same measure and a concentration that is toxic for on organ is not probably toxic for other organs. Finally, it seems that exposure to a high concentration of zinc nano-oxide compared to same concentration but in divided doses and longer time has more destructive effects on body tissues.

**References**

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