

# In-situ compaction and sintering of Al<sub>2</sub>O<sub>3</sub>-B<sub>4</sub>C composites by using a High-Frequency Induction System

U. Çavdar<sup>1\*</sup>, A. Altıntaş<sup>2</sup>, B. Karaca<sup>2</sup>

<sup>1</sup>*Izmir Demokrasi University, Engineering Faculty, Mechanical Engineering Department, IDU Campus, 35140, Izmir, Turkey*

<sup>2</sup>*Manisa Celal Bayar University, Mechanical Engineering Department, Muradiye Campus Manisa, Turkey*

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## Abstract

Alumina powders containing varying ratios of B<sub>4</sub>C were mixed in a V-type mixer for 30 min with 35 cycle/min to obtain a homogeneous mixture. Six different chemical compositions were prepared and sintered at 1600 °C for 20 min using an Ultra-High-Frequency Induction Heated System. The samples were cooled by air. The density, surface roughness, hardness and microstructure of samples were investigated. The results showed that when the amount of B<sub>4</sub>C increased, the values of hardness increased and the values of surface roughness decreased.

**Key words:** alumina, boron carbide, sintering, induction, composite

## 1. Introduction

Comparing powder metallurgy (PM) with other conventional manufacturing technologies, machining process is not necessary requirement, as standard parts with complex geometry in a wide composition range can be manufactured serially. This production method has many advantages, such as low investment costs, the equipment used is flexible, high productivity, and it is easy to produce materials with different properties [1–3].

Alumina oxide is a chemical composition made up of two units of aluminium and three units of oxygen with the chemical formula Al<sub>2</sub>O<sub>3</sub>. Aluminium oxide is one of the most used materials in the industry because of its superior hardness, chemical inertness and electrical/thermal insulation properties [4]. Alumina is used in ceramic, glass, refractory industry, grinder material production, primary aluminium production, transparent armour production and ceramic cutting production [5–7].

Boron carbide (chemical formula B<sub>4</sub>C) is generally used in tank armour and engine sabotage powders in industrial applications [8]. It is a highly durable product. The melting temperature, boiling point and Young's modulus of B<sub>4</sub>C are 2450 and 3500 °C and

450–470 GPa, respectively [9–11].

Sintering is a heating process applied to a powder compact to strengthen and bond it. The temperature used for sintering is below the melting temperature of the major constituent of the Powder Metallurgy (PM) material [12, 13]. Spark Plasma Sintering (SPS) is similar to microwave sintering, inducing fast sintering due to its own thermal environment. Generally, this method is performed on Al<sub>2</sub>O<sub>3</sub>, SiC and ZrO<sub>2</sub> [14, 15]. In the induction sintering system, compacts are heated very quickly. The High-Frequency Induction Heated Sintering (HFIHS) process is similar to SPS. This method is usually applied to ceramic materials [16].

Al<sub>2</sub>O<sub>3</sub>-14wt.%B<sub>4</sub>C composites with 0-1wt.%C addition were sintered in an Ar atmosphere at 1550–1650 °C by Huang et al. The densification behaviour and microstructure of the Al<sub>2</sub>O<sub>3</sub>-B<sub>4</sub>C composites were investigated [17].

Al<sub>2</sub>O<sub>3</sub>-TiC-B<sub>4</sub>C powders were mixed in different proportions by Sun et al. A low-pressure hot pressing process was applied to all mixtures, then the mechanical properties of all the compacts were investigated. The Vickers hardness, fracture toughness and flexural strength of the composite with the addition of 4.7wt.%Al<sub>2</sub>O<sub>3</sub> and 10wt.%TiC, were 24.8 GPa,

\*Corresponding author: tel.: +90 232 260 1001; fax: +90 232 260 1004; e-mail address: [ugur.cavdar@idu.edu.tr](mailto:ugur.cavdar@idu.edu.tr)

Table 1. Chemical composition of Al<sub>2</sub>O<sub>3</sub> and B<sub>4</sub>C powders (wt.%)

Material	Al <sub>2</sub> O <sub>3</sub>	N <sub>2</sub> O (max)	B	C	Fe <sub>2</sub> O <sub>3</sub> (max)	SiO <sub>2</sub> (max)	Free B	CaO (max)	Free C	MgO (max)	TiO <sub>2</sub> (max)
Al <sub>2</sub> O <sub>3</sub>	99.73	0.14	–	–	0.03	0.01	–	0.02	–	0.01	0.02
B <sub>4</sub> C	–	–	78	20.13	0.17	–	0.1	–	0.92	–	–

Table 2. The codes and chemical compositions of specimens

Code	Chemical compositions of specimens
B0	0 wt.% Boron carbide + 100 wt.% Pure alumina
B1	1 wt.% Boron carbide + 99 wt.% Pure alumina
B5	5 wt.% Boron carbide + 95 wt.% Pure alumina
B10	10 wt.% Boron carbide + 90 wt.% Pure alumina
B30	30 wt.% Boron carbide + 70 wt.% Pure alumina
B50	50 wt.% Boron carbide + 50 wt.% Pure alumina

Table 3. Processing parameters of a high-frequency induction heated compacts

Parameters	Applied value for induction sintering
Maximum temperature	1600 °C
Power capacity	2.5 kW
Frequency	900 kHz
Pre-sintering time	Unavailable
Duration	30 s
Heating rate	~ 20 °C s <sup>-1</sup>
Cooling rate	Naturally
Environment	Atmosphere

4.8 MPa m<sup>1/2</sup> and 445 MPa, respectively [18].

Al<sub>2</sub>O<sub>3</sub>-B<sub>4</sub>C powders were sintered successfully with no pressure. The mechanical properties of the Al<sub>2</sub>O<sub>3</sub>-B<sub>4</sub>C compacts were investigated. The results showed that when the amount of B<sub>4</sub>C increased, the density of the composites decreased.

## 2. Experimental method

Alumina (particle size of 2.3 μm, purity of 99.4 %) was used as the starting material. Boron carbide (B<sub>4</sub>C) (average particle diameter of 63 μm, purity of 99.6 %) was supplied by Boroptik Engineering, R&D Manufacturing and Trade Co. The chemical composition of Al<sub>2</sub>O<sub>3</sub> and B<sub>4</sub>C is shown in Table 1.

All the powders' compositions, which are shown in Table 2 were mixed in a V-type mixer for 30 min at 35 cycles/min to obtain a homogeneous mixture. The size of the Al<sub>2</sub>O<sub>3</sub>-B<sub>4</sub>C samples was 10 × 3 mm<sup>2</sup> and they weighed approximately 0.5 g.

Table 4. Density values of all compositions (Error Range [ER] is between ± 1 μm)

Code	Density (g cm <sup>-3</sup> )
B0	3.476
B1	3.362
B5	3.319
B10	3.271
B30	2.985
B50	2.622

The mixtures of Al<sub>2</sub>O<sub>3</sub>-B<sub>4</sub>C obtained were placed in the graphite die (outside diameter 45 mm; inside diameter 20 mm; height 18 mm). The induction sintering process was carried out using a 2.5 kW power supply at a frequency of 900 kHz. All compacts were sintered at 1600 °C for 20 min dwell time with an applied pressure of 30 MPa. Duana et al. [19] compacted their powders using the same sintering pressure. In contrast to Çavdar [20] and Çavdar and Atik [16] who used 5 minutes dwell time for induction sintering of the metallic materials, in this work a sintering dwell time of 20 min was used due to the sintering application of the ceramic composition. The temperature of the compacts was measured by an infrared thermometer (± 5 °C) throughout the induction sintering process. After the completion of the sintering process, all compacts were cooled by air. All the sintering parameters are given in Table 3.

The relative densities of compacts were evaluated by the Archimedes' Method. Radwag AS 220/C/2 Archimedes' Scale was used for this test. Surface roughness was measured with the Mitutoyo surf test SJ-301 Surface Test Device. After the surface roughness test, an abrasion test was conducted on the specimens. In this study, a WC ball was used and applied at a normal load of 10 N with a sliding speed of 20 cm s<sup>-1</sup> for 1000 m. The weight loss of the specimens was calculated and the results are given in Table 5.

The surface hardness was evaluated using the FM 700 Micro-hardness Tester with an automatic load of 31.25 kg and a dwell time of 12 s. The values of Vickers hardness were obtained. The Metkon Forcipol IV was used for etching and polishing. Micro-structural characterisation was made on the product samples which had been polished and etched using a solution of

Table 5. Average surface roughness ( $\mu\text{m}$ ) (ER is between  $\pm 1 \mu\text{m}$ ) and weight loss (%) (ER is between  $\pm 0.5 \%$ ) of sintered samples

Code	$R_a$	$R_y$	$R_z$	Weight loss
B0	2.08	21.13	14.15	-2.06
B1	2.37	21.48	14.59	-1.99
B5	2.62	21.84	14.94	-1.62
B10	3.27	22.45	16.08	-0.99
B30	4.01	24.17	17.30	+0.32
B50	5.12	25.42	18.64	+0.41

$\text{C}_2\text{H}_6\text{O}$  (97 vol.%),  $\text{HNO}_3$  (3 vol.%). The microstructures were observed using a MEIJO metal microscope and the images obtained are shown in Figs. 2, 3.

### 3. Results and discussion

The relative densities of all compacts were measured by using the Archimedes' method. The theoretical density of  $\text{Al}_2\text{O}_3$  and  $\text{B}_4\text{C}$  is  $3.95$  and  $2.51 \text{ g cm}^{-3}$ . It was observed that density values changed between  $3.476$  and  $2.62276 \text{ g cm}^{-3}$  as shown in Table 4, and the percentage of density values changed from 88 to 81.2%. Sun et al. [21] obtained a higher concentration of  $\text{Al}_2\text{O}_3$ - $\text{B}_4\text{C}$  composite as the sintering process was carried out at a higher temperature. The results show that the relative density of the specimens rapidly decreased when the amount of  $\text{B}_4\text{C}$  increased. The melting temperature of alumina and  $\text{Al}_2\text{O}_3$  is  $2072$  and  $2500^\circ\text{C}$ . Alumina was sintered better than  $\text{B}_4\text{C}$  at  $1600^\circ\text{C}$ . Thus, the highest density was obtained with pure alumina.

The surface roughness test was applied to all sintered compacts. The surface roughness values are shown in Table 5. In the same table,  $R_a$  is the arithmetic average of the absolute values,  $R_y$  means the highest micro-peak to the lowest micro-trough vertical distance within a single sample length,  $R_z$  is a parameter that averages the height of the five highest micro-peaks plus the depth of the five deepest micro-troughs over the evaluation length [22]. According to the results, when the amount of  $\text{B}_4\text{C}$  increased, surface roughness values increased. This is because the grain size of  $\text{B}_4\text{C}$  is bigger than alumina's and hence the porosity of the material is greater.

The wear test was applied to the samples. Before and after the wear test, their weights were measured. Percentage weight loss was calculated and is given in Table 5. As shown in the table, when the amount of  $\text{B}_4\text{C}$  increased, the percentage weight loss decreased. For the 30 and 50wt.% $\text{B}_4\text{C}$  reinforced compacts, the weight of the specimens increased. Due to the extreme increase in hardness and brittleness of both compacts,

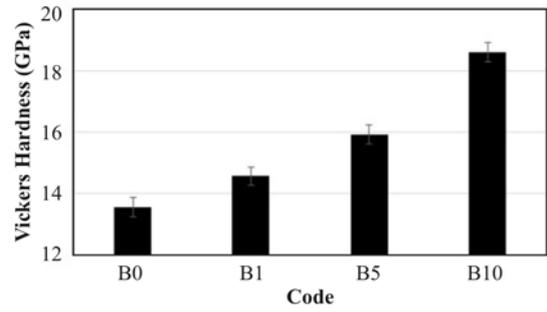


Fig. 1. The average hardness of sintered composites.

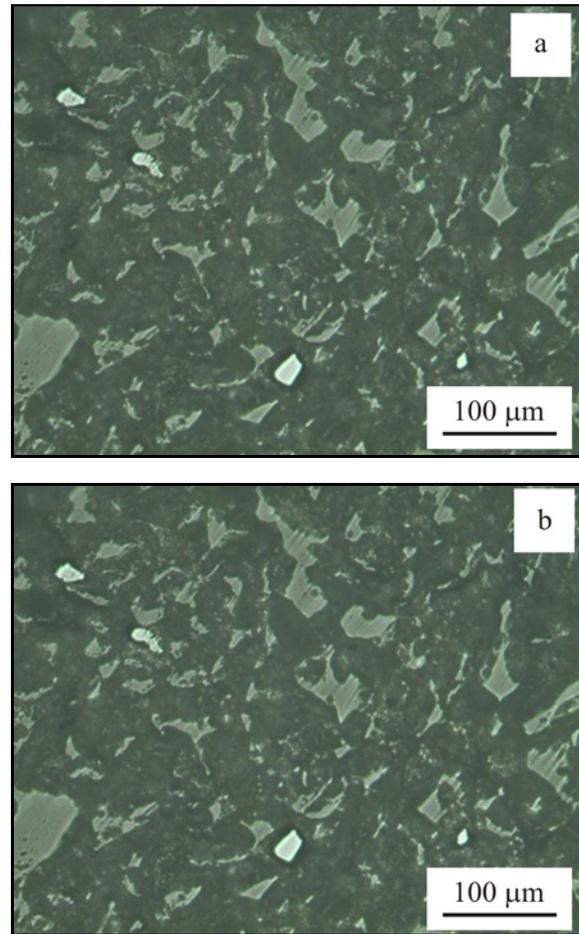


Fig. 2. (a) 1wt.% $\text{B}_4\text{C}$ -99 wt.%  $\text{Al}_2\text{O}_3$  and (b) 5wt.% $\text{B}_4\text{C}$ -95wt.% $\text{Al}_2\text{O}_3$  compacts were taken from the centre of the surface of the microstructural images.

abrasion was seen on the ceramic balls' counter-bodies for these two compositions. These results demonstrate that the hardness of the compacts is higher than the hardness of the ceramic balls.

Vickers hardness was taken from five different points and the average values calculated are given in Fig. 1. According to results, when the amount of  $\text{B}_4\text{C}$  increased, the hardness of sintered compacts in-

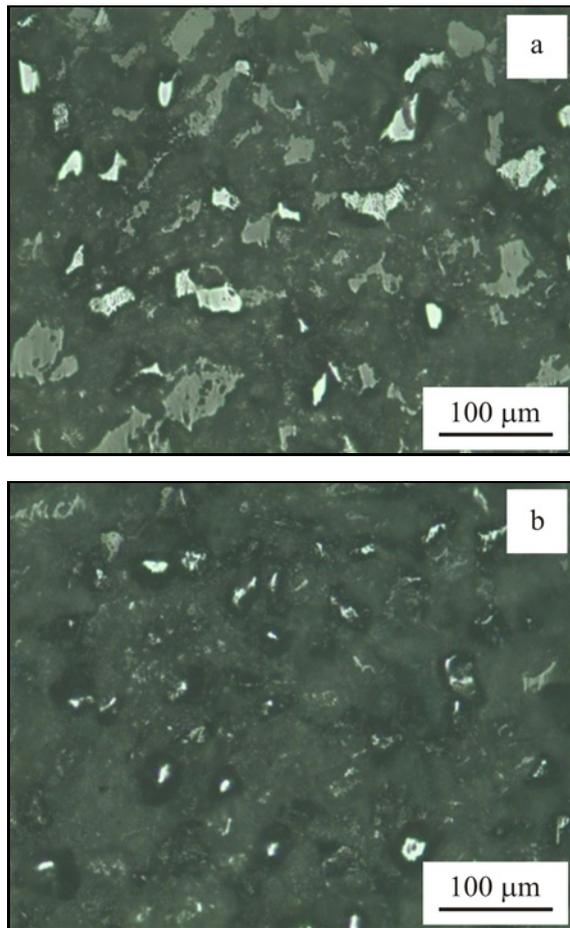


Fig. 3. (a) 10wt.%B<sub>4</sub>C-90wt.%Al<sub>2</sub>O<sub>3</sub> and (b) 30wt.%B<sub>4</sub>C-70wt.%Al<sub>2</sub>O<sub>3</sub> compacts were taken near the broken surface of microstructural images.

creased. This is because the hardness of B<sub>4</sub>C is higher than that of the alumina. Also, when the amount of B<sub>4</sub>C increased, the sintering temperature of alumina and boron carbide changed which showed us that we had to consider the alumina and boron carbide sintering temperature, too. The sintering temperature of alumina is less than that of B<sub>4</sub>C. The hardness values of 70wt.%Al<sub>2</sub>O<sub>3</sub>-30wt.%B<sub>4</sub>C (B30) and 50wt.%Al<sub>2</sub>O<sub>3</sub>-50wt.%B<sub>4</sub>C (B50) compacts could not be measured due to the formation of weak bonds between the grains during the sintering process, and hence the compacts broke. Sun et al. [23] investigated the effects of alumina addition to boron carbide ceramics prepared by the SPS technique. They used sintering temperatures from 1700 to 1800 °C. In contrast to this study, we sintered our compacts at 1600 °C. Consequently, the broken B30 and B50 compacts indicate that our sintering temperature was inadequate for boron carbide reinforcements above 10 wt.%.

The microstructures of grain boundaries of sintered pellets are shown in Figs. 2, 3. It was seen that addi-

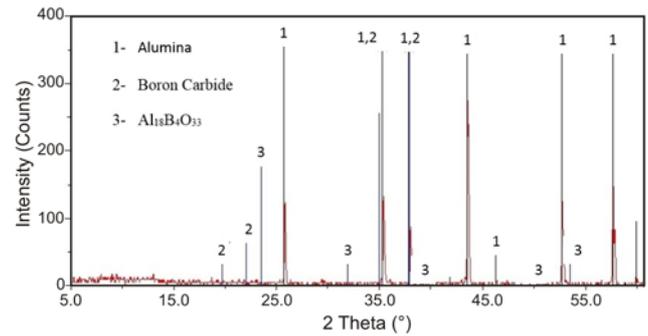


Fig. 4. XRD patterns of Al<sub>2</sub>O<sub>3</sub>-B<sub>4</sub>C composite.

tive material of B<sub>4</sub>C is dispersed homogeneously in the composite. It is also seen that the denser the mixture, B<sub>4</sub>C dispersed more uniformly in Al<sub>2</sub>O<sub>3</sub> 30 wt.% and 50 wt.% B<sub>4</sub>C containing Al<sub>2</sub>O<sub>3</sub> compacts, shown in the images taken near the broken surface of the microscope image (Figs. 3a,b). The rate of porosity increased, and weak bonds between grains occurred during the sintering process due to increasing the amount of B<sub>4</sub>C. For this reason, the composite had a more brittle structure.

In-situ observation of the dark scattered images revealed that the microstructures of Al<sub>2</sub>O<sub>3</sub> ceramics contained a white B<sub>4</sub>C matrix with a dispersed grey phase as shown in Figs. 2, 3. Based on XRD analysis, the grey phase was identified as Al<sub>2</sub>O<sub>3</sub>-B<sub>4</sub>C ceramics.

Figure 4 summarises the results of the XRD phase identification of ultra-high frequency induction sintered Al<sub>2</sub>O<sub>3</sub>-B<sub>4</sub>C ceramic. Previous reports confirmed the formation temperature (about 1100 °C) of Al<sub>18</sub>B<sub>4</sub>O<sub>33</sub> phase by reaction (Eq. (1)) [24, 25]. After the sintering process, these studies detected the same Al<sub>18</sub>B<sub>4</sub>O<sub>33</sub> phase in our composition. During sintering, the reaction occurred between Al<sub>2</sub>O<sub>3</sub> and B<sub>4</sub>C. This reaction is given by Eq. (1):



#### 4. Conclusions

In this study, the mechanical properties of high-frequency induction heated sintered B<sub>4</sub>C powder of different weights were studied. The effect of B<sub>4</sub>C on alumina was investigated. The following results were obtained:

- The surface roughness value reached a maximum level at 50wt.%B<sub>4</sub>C-50Al<sub>2</sub>O<sub>3</sub> compacts.
- Highest density was obtained for pure alumina and highest hardness result was obtained for Al<sub>2</sub>O<sub>3</sub>-10wt.%B<sub>4</sub>C.
- As the amount of B<sub>4</sub>C increased, the compos-

ite structure became more brittle. Hence, fracturing occurred during the measurement of hardness.

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