

## VITAMIN D AND ITS METABOLITES – SUPPLY OF PATIENTS WITH VARIOUS ENDOCRINE DISORDERS AND COMPARISON OF ANALYTICAL METHODS

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**Objective.** Vitamin D is important not only for its effect on the homeostasis of calcium, but also for its anti-proliferative, pro-differentiation, pro-apoptotic and immuno-modulating effects in the tissues of living organisms. This work describes the results of serum 25(OH)D (n=2175) and 1 $\alpha$ ,25(OH)<sub>2</sub>D (n=2271) radioimmunoassays (RIA) conducted at the Institute of Endocrinology in Prague (Czech Republic) during the period of 2004-2006.

**Methods.** Serum concentrations of vitamin D metabolites were determined by RIA kits from IDS Ltd., Boldon, UK. In a group of 20 healthy volunteers, results obtained using the RIA kit were compared with those obtained after serum extraction with acetonitrile, subsequent chromatographic (RP HPLC) separation of 25(OH)D and its detection in collected fractions using the same RIA kit from IDS.

**Results.** The mean concentrations  $\pm$  S.D. in the samples studied were 76.4  $\pm$  45.6 nmol/l for 25(OH)D and 88.9  $\pm$  45.1 pmol/l for 1 $\alpha$ ,25(OH)<sub>2</sub>D. The mean concentrations of both metabolites were higher in women than in men, but only the difference for 1 $\alpha$ ,25(OH)<sub>2</sub>D was statistically significant (p=0.0000). The lowest concentrations of 25(OH)D or 1 $\alpha$ ,25(OH)<sub>2</sub>D were found in patients with hyperparathyroidism, the highest concentrations those treated with cholecalciferol for osteoporosis and vitamin D deficiency. The relationship between 25(OH)D and 1 $\alpha$ ,25(OH)<sub>2</sub>D was expressed by the equation 1 $\alpha$ ,25(OH)<sub>2</sub>D = 71.0845 + 0.1890 \* 25(OH)D (n=1065, p=0.0000, R<sup>2</sup>=0.0343). Based on the RIA results, only 4 % of individuals were inadequately supplied with 25(OH)D, 79 % were supplied adequately and almost 17 % were supplied more than adequately. Similarly, 1 $\alpha$ ,25(OH)<sub>2</sub>D shows inadequate concentration in 15 % of the individuals tested, 59 % of these individuals were within the reference range and 26 % had values exceeding the upper limit of the reference range. The mean concentrations of 25(OH)D obtained by HPLC separation were in average by 8 % higher than those obtained using only the RIA kit.

**Conclusion.** Interpretation of studies dealing with vitamin D stores must be based on precise and correct analytical procedures. However, only a few information exists about the influence of other vitamin D metabolites and conjugates on the concentration of immunoanalytically measured serum 25(OH)D or 1 $\alpha$ ,25(OH)<sub>2</sub>D.

**Key words:** 1 $\alpha$ ,25-dihydroxyvitamin D – 25-hydroxyvitamin D – radioimmunoassay – serum – chromatography – comparison of methods

Vitamin D is acquired by the organism exogenously in the form of vitamin D<sub>2</sub> (ergocalciferol) which is present in plants and fungi, but the main source of vitamin D is its endogenous biosynthesis from the pre-

cursor – 7-dehydrocholesterol. After the exposure of skin to sunlight at wavelengths of 290-315 nm, the precursor undergoes photolysis which results in the opening of ring B. Spontaneous isomeration of such inter-

mediate then results in a synthesis of vitamin D<sub>3</sub> (cholecalciferol) (WHITING and CALVO 2005). The average values of serum 25(OH)D<sub>3</sub> (20.5 ng/ml) determined by liquid chromatography/mass spectrometry exceeded by approximately 50 times the concentration of 25(OH)D<sub>2</sub> (0.4 ng/ml) (TSUGAWA et al. 2005). Important factors that play a role in supplying the organism by vitamin D<sub>3</sub> include, besides the sunlight, also the area of skin accessible to ultraviolet radiation, skin pigmentation and the age of the individuals (BOUILLON et al. 2005a). From the view of biological function, the question remains whether vitamin D<sub>2</sub> is equivalent to vitamin D<sub>3</sub>. TRANG et al. (1998) stated that vitamin D<sub>2</sub> increases the level of 25-hydroxyvitamin D (25(OH)D) less effectively than D<sub>3</sub>, and that the nutritional effect of both vitamins D is not equivalent. Vitamins D<sub>2</sub> and D<sub>3</sub> differ in the structure of the side chain of their molecule at the 17 $\alpha$ -position of the secosteroid (vitamin D<sub>2</sub> – has one additional double bond and one additional methyl group). However, their subsequent metabolic pathway is identical (SHIMADA et al. 2003). Under the influence of hydroxylases from the cytochrome P-450 enzyme family, vitamin D is first hydroxylated in the C25 position via 25-hydroxylase (CYP27A) produced by the liver. This gives rise to 25(OH)D. The concentration of this biologically as yet inactive metabolite of vitamin D in blood reflects the amount of vitamin D stores in the organism from exogenous (D<sub>2</sub>, D<sub>3</sub>) as well as endogenous (D<sub>3</sub>) sources (BOUILLON et al. 2005a; TURPEINEN et al. 2003). The biologically active and hormonally effective 1 $\alpha$ ,25-dihydroxyvitamin D (1 $\alpha$ ,25(OH)<sub>2</sub>D) is formed in the kidney by the hormonally regulated 1 $\alpha$ -hydroxylase (CYP27B) which is stimulated by parathyroid hormone (PTH), while 1 $\alpha$ ,25(OH)<sub>2</sub>D has a negative feedback effect. The kidneys, as well as other tissues, contain the enzyme 24-hydroxylase (CYP24), which catalyses the synthesis of (24R)-24,25(OH)<sub>2</sub>D from 25(OH)D, but mainly inactivates 1 $\alpha$ ,25(OH)<sub>2</sub>D by transforming it to 1 $\alpha$ ,24,25(OH)<sub>3</sub>D (SHIMADA et al. 2003). Overall, 37 various types of vitamin D<sub>3</sub> metabolites have been isolated and chemically characterized (<http://vitamind.ucr.edu/biochem.html>, access 7.1.2006).

1 $\alpha$ ,25(OH)<sub>2</sub>D is a key regulator of calcium and phosphorus homeostasis, inducing the absorption of gastrointestinal calcium and mobilizing calcium from bone. It also exerts strong anti-proliferative, pro-differentiation, pro-apoptotic and immuno-modulatory (suppression of humoral response, activation of cellular response) effects (BOUILLON et al. 2005a; VAN ETTEN and MATHIEU 2005; BOUILLON et al. 2005b, STEWART et al.

2005). Long-term vitamin D deficiency and consequently decreased 1 $\alpha$ ,25(OH)<sub>2</sub>D levels leads to rickets in children and to osteomalacia in adults. Together with calcium deficiency and other metabolic changes related to increasing age, vitamin D deficiency plays a role in the pathogenesis of osteoporosis (FRASER et al. 1997), alopecia (HSIEH et al. 2003), muscle weakness, development of secondary hyperparathyroidism in patients with chronic renal insufficiency.

Current research shows that vitamin D deficiency may be associated with hypertension, depression and immune system disorders that it also leads to multiple sclerosis, rheumatic arthritis and diabetes (Siu-Caldera et al. 1999). There is an increasing number of studies showing a relationship between the prevalence of cancer, especially of the colon, prostate and breast as well as other malignancies on one side and the amount of vitamin D body stores on the other (Bouillon et al. 2005a).

The concentration of 25(OH)D in blood is the best marker of vitamin D store (WHITING and CALVO 2005, BOUILLON et al. 2005a, QUESADA et al. 2004, MATA-GRANADOS et al. 2004). Within routine clinical examinations, serum 25(OH)D levels are usually determined using commercial immunoassays (radioimmunoassay (HOLLIS 2000), ELISA (LIND et al. 1997)) with or without chromatographic or extraction purification. The advantage of immunoassays is the large capacity and the analytical sensitivity in the order of pg per test tube. Immunoassays of the hydroxylated metabolites of vitamin D are also affected by a number of problems. One such problem involves the strong binding of vitamin D metabolites to the vitamin D binding protein (VDBP) and to a lesser extent to albumin during RIA (radioimmunoassay) quantification. Serum precipitation using organic solvents or acids may thus lead to losses due to co-precipitation of vitamin D metabolites with the binding proteins (QUESADA et al. 2004). Another difficulty involves the standardisation of the immunoanalytical kits used. Various types of assays yield various results from which various reference ranges are derived and clinical decisions created. These do not necessarily reflect the true vitamin D stores within a given population (GLENDEENING 2003). Methods based on the competition between specific binding proteins and a mixture of labeled (tracer) and unlabeled (assayed) ligands lack adequate specificity to differentiate between vitamin D<sub>2</sub> and vitamin D<sub>3</sub> metabolites. From this point of view the error in RIA determination of vitamin D metabolites can depend on the applied

system and on the proportion of individual types of vitamin D metabolites in the circulation.

Another key issue is the antibody specificity when determining  $1\alpha,25(\text{OH})_2\text{D}$ , whose concentration is lower than that of  $25(\text{OH})\text{D}$  by approximately an order of 3. Inadequately effective processes of separation of  $1\alpha,25(\text{OH})_2\text{D}$  from  $25(\text{OH})\text{D}$ , or the use of an antibody with even minimum cross-reactivity with  $25(\text{OH})\text{D}$  may lead to false high results related to  $1\alpha,25(\text{OH})_2\text{D}$ . The specificity of used antibodies against the conjugates (sulphate, glucuronide) of hydroxylated vitamin D metabolites is not being discussed at all. However, the determination of plasma  $25(\text{OH})\text{D}_3$  and  $25(\text{OH})\text{D}_3$ -3-sulphate using reverse-phase high-pressure liquid chromatography (RP HPLC) and ultraviolet (UV) detection at 265 nm has shown that the plasma of healthy persons contains a nearly equivalent amount of both substances ( $n=15$ , 15 ng/ml  $25(\text{OH})\text{D}_3$ -3-sulphate, 14 ng/ml  $25(\text{OH})\text{D}_3$ ) (SHIMADA et al. 1997, SHIMADA et al. 1995). Similar concentrations 16.7 ng/ml for  $25(\text{OH})\text{D}_3$ -3-sulphate were obtained by gas chromatography/mass spectrometry (AXELSON 1985).

This work describes the results obtained by radioimmunoassays of two main metabolites of vitamin D, i.e.  $25(\text{OH})\text{D}$  and  $1\alpha,25(\text{OH})_2\text{D}$  in a group of patients of the Institute of Endocrinology (Prague, Czech Republic) in 2004-2006. We developed a simple RP HPLC separation of  $25(\text{OH})\text{D}$  following prior extraction of  $25(\text{OH})\text{D}$  from serum with acetonitrile. Parallel determination of  $25(\text{OH})\text{D}$  by commercially available RIA kit or following RP HPLC separation of  $25(\text{OH})\text{D}$  from serum with the same RIA detection served to comparison of results. The RP HPLC procedure is also suitable for the determination levels of other lipophilic vitamins (e.g. A, E, K1) assuming that more effective detection is used than that involving a UV detector at 254 nm or 290 nm.

## Materials and Methods

**Chemicals.** All chemicals used were of analytical grade. Vitamins  $\text{D}_3$ ,  $\text{D}_2$  and their metabolites  $25(\text{OH})\text{D}_3$ ,  $25(\text{OH})\text{D}_2$ ,  $1\alpha,25(\text{OH})_2\text{D}$ , as well as ((±)-(-)-tocopherol (vit. E), all trans-retinal (vit. A) and vitamin  $\text{K}_1$  were purchased from Sigma-Aldrich (Stenheim, Germany), labeled  $^{125}\text{I}$ - $25(\text{OH})\text{D}_3$  from IDS (Baldon, UK); labeled  $^3\text{H}$ - $25(\text{OH})\text{D}_3$  from Perkin Elmer Life Sciences (Boston, USA). We used methanol and acetonitrile from Merck (Darmstadt, Germany); diethylether, dichloromethane and n-hexane from Lachema (Brno, Czech

Republic). Deionised water was prepared by a water-treatment apparatus from Millipore AFS-18D (Pittsburgh, USA).

Stock solutions of vitamins and their metabolites at concentrations of 10 µg /100 µl were prepared by dissolving the given analyte in acetonitrile. These solutions were stored in darkness at -18 °C.

For the conversion of units following equations were used:  $1\alpha,25(\text{OH})_2\text{D}_3$  (pmol/l)  $\times$  0,417 =  $1\alpha,25(\text{OH})_2\text{D}_3$  (pg/ml);  $25(\text{OH})\text{D}_3$  (nmol/l)  $\times$  0,401 =  $25(\text{OH})\text{D}_3$  (ng/ml);  $1\alpha,25(\text{OH})_2\text{D}_2$  (pmol/l)  $\times$  0,429 =  $1\alpha,25(\text{OH})_2\text{D}_2$  (pg/ml);  $25(\text{OH})\text{D}_2$  (nmol/l)  $\times$  0,417 =  $25(\text{OH})\text{D}_2$  (ng/ml).

**Patients.** During the period of 2004 – 2006, serum concentrations of  $25(\text{OH})\text{D}$  ( $n=2175$ , 1937 women, 238 men) and  $1\alpha,25(\text{OH})_2\text{D}$  ( $n=2271$ , 2024 women, 247 men) were determined by RIA kits in the Institute of Endocrinology. The subjects were 1-93 years old (mean  $\pm$  SD =  $56 \pm 18$ ) (Table 1). Hyperparathyroidism was diagnosed in 259 patients, osteoporosis or vitamin D deficiency in 336 patients, defective function of thyroid gland in 2332 patients. Various endocrine disturbances were found in 454 patients. Parallel determination of  $25(\text{OH})\text{D}$  together with  $1\alpha,25(\text{OH})_2\text{D}$  was made in 1065 patients (928 women, 137 men).

**Radioimmunoassay of  $25(\text{OH})\text{D}$  and  $1\alpha,25(\text{OH})_2\text{D}$ .** RIA IDS Gamma-B25-D-hydroxy-vitamin D immuno-diagnostic kit from IDS (Baldon,UK) was used for the determination of  $25(\text{OH})\text{D}$  in serum aliquots of 50 µl. In accordance with the information provided by manufacturer, the reference range for such kit is 23-113 nmol/l, an analytical sensitivity less than 3 nmol/l and a specificity (cross-reactions in %) of sheep polyclonal antibody 100 % for  $25(\text{OH})\text{D}_3$ , 75 % for  $25(\text{OH})\text{D}_2$ , more than 100 % for  $24,25(\text{OH})_2\text{D}_3$ , and less than 0.01 % or 0.3 % for  $\text{D}_3$ , and  $\text{D}_2$  respectively.

$1\alpha,25(\text{OH})_2\text{D}$  was determined in serum aliquots of 500 µl using the immuno-extraction Gamma-B1,25-Dihydroxy Vitamin D RIA kit from the same company. In this case the immuno-extraction is conducted with the aid of monoclonal antibodies immobilized to solid particles. The RIA itself uses a polyclonal sheep antibody whose analytical sensitivity is 8 pmol/l and which shows 100 % specificity for  $1\alpha,25(\text{OH})_2\text{D}_3$ , 91 % for  $1\alpha,25(\text{OH})_2\text{D}_2$ , less than 0.01 % for  $24,25(\text{OH})_2\text{D}_3$  and less than 0.001 % for  $25(\text{OH})\text{D}_3$ . The reference range of  $1\alpha,25(\text{OH})_2\text{D}$  is 48-110 pmol/l.

**RP HPLC. A. Extraction of vitamin D metabolites from serum.** 2000 µl of acetonitrile were added to 500 µl of serum. After vortexing for 10 s the mixture was centrifuged in a cooled centrifuge (4 °C) for 10

min at 2000 g. The supernatant was evaporated using a vacuum centrifuge (2 h at 40 °C) and the vaporized material was stored at -18 °C in polypropylene test tubes until analysis. Before the determination, the vaporized material was reconstituted in acetonitrile. The effectiveness of extraction was evaluated by the addition of  $^3\text{H}$ -25(OH) $_2\text{D}_3$  to the analyzed serum and the measurement of radioactivity in the extract.

**B. Analysis with UV detection of vitamin D metabolites and RIA detection of 25(OH) $_2\text{D}$ .** A liquid chromatograph was used (Laboratorní přístroje Prague, Czech Republic) including: 2 x HPP 5001 pump, GP2 gradient controls, LCI 30 dispenser, UV filter detector 254 nm LCD 2563, UV grid detector 290 nm LCD 2040, steel column (250 x 4,6 mm i.d.) filled with the stationary phase Separon SGX RPC18 (dp7  $\mu\text{m}$ ) from Tessek (Czech Republic). The Apex Integrator 3.1 (DataApex, Czech Republic) served to evaluate the chromatograms.

The chromatography column was washed prior to each analysis with methanol and acetonitrile, than was equilibrated for 5 minutes with the mobile phase. Vitamin D metabolites (injection of 100  $\mu\text{l}$  vaporized serum extract reconstituted in acetonitrile) were separated isocratically by the mobile phase composed of methanol-acetonitrile = 95 - 5, at a flow rate of 1.5 ml/min. The detection was performed by UV detectors working at wavelengths of 254 nm and 290 nm.

The individual fractions of the eluate corresponding to the retention time of 25(OH) $_2\text{D}$  ( $t_r$  = 3.0 - 4.5 min.) were collected in 0.5 minute intervals (3 x 750  $\mu\text{l}$ ), drained off and vaporised within one hour at 40 °C using the vacuum centrifuge RC 10.10 (Jouan, France). The vaporized material was reconstituted in 800  $\mu\text{l}$  of

acetonitrile and 50  $\mu\text{l}$  of the solution was used for the IDS immunoassay of 25(OH) $_2\text{D}$ .

**Statistical evaluation.** All statistical procedures were conducted using the NCSS 2001 program (Hintze 2006) at a probability level of 0.05. In view of the non-normal distribution, we used non-parametric tests (Mann-Whitney U test and Wilcoxon rank-sum test for difference in medians, Kolgomorov-Smirnov test for normal distributions) as well as parametric tests (t-test for equivalent and non-equivalent distribution) to evaluate the statistical significance of value differences.

## Results

**Radioimmunoassay data.** The average values of RIA assays of serum 25(OH) $_2\text{D}$ , 1 $\alpha$ ,25(OH) $_2\text{D}$  are listed in Table 1. The mean and median serum levels of 25(OH) $_2\text{D}$  (73.8 nmol/l in men, 76.8 nmol/l in women) and 1 $\alpha$ ,25(OH) $_2\text{D}$  (76.0 pmol/l in men, 90.5 pmol/l in women) were higher in women than in men (Table 1). But only in the case of 1 $\alpha$ ,25(OH) $_2\text{D}$  the difference was statistically significant using both parametric and non-parametric tests ( $p < 0.0000$ ). The distribution of 25(OH) $_2\text{D}$  and 1 $\alpha$ ,25(OH) $_2\text{D}$  values is shown in Fig. 1 and the relationship between 25(OH) $_2\text{D}$  and 1 $\alpha$ ,25(OH) $_2\text{D}$  is shown in Fig. 2. This relationship is described on the basis of linear regression by the equation  $1\alpha,25(\text{OH})_2\text{D} = 71.0845 + 0.1890 * 25(\text{OH})_2\text{D}$ , and is highly significant ( $p < 0.0000$ ) with the coefficient of determination  $R^2 = 0.0343$ , Pearson's correlation coefficient of 0.1853 and Spearman's correlation coefficient of 0.2223. Fig. 3 represents seasonal changes of 25(OH) $_2\text{D}$  and 1 $\alpha$ ,25(OH) $_2\text{D}$  concentrations as average values obtained in the individual months. Based on the

**Table 1**  
Mean concentrations of serum 25(OH) $_2\text{D}$  and 1 $\alpha$ ,25(OH) $_2\text{D}$ .

	n	age (years) $\pm$ SD	mean	stds	median	range	reference range
25(OH) $_2\text{D}$ (nmol/l)							
Total	2175	57.5 +/- 16.9	76.4	45.6	67.7	7.5 - 406.2	23-113
Men	238	47.3 +/- 21.5	73.8	51.8	60.5	8.8 - 406.2	23-113
Women	1937	58.7 +/- 15.8	76.8	44.8	68.3	7.5 - 380.2	23-113
1 $\alpha$ ,25(OH) $_2\text{D}$ (pmol/l)							
Total	2271	55.4 +/- 18.9	88.9	45.1	85.7	1.3 - 514.8	48-110
Men	247	51.4 +/- 19.8	76.0	49.1	75.3	2 - 258.1	48-110
Women	2024	55.9 +/- 18.7	90.5	44.3	87.4	1.3 - 514.8	48-110

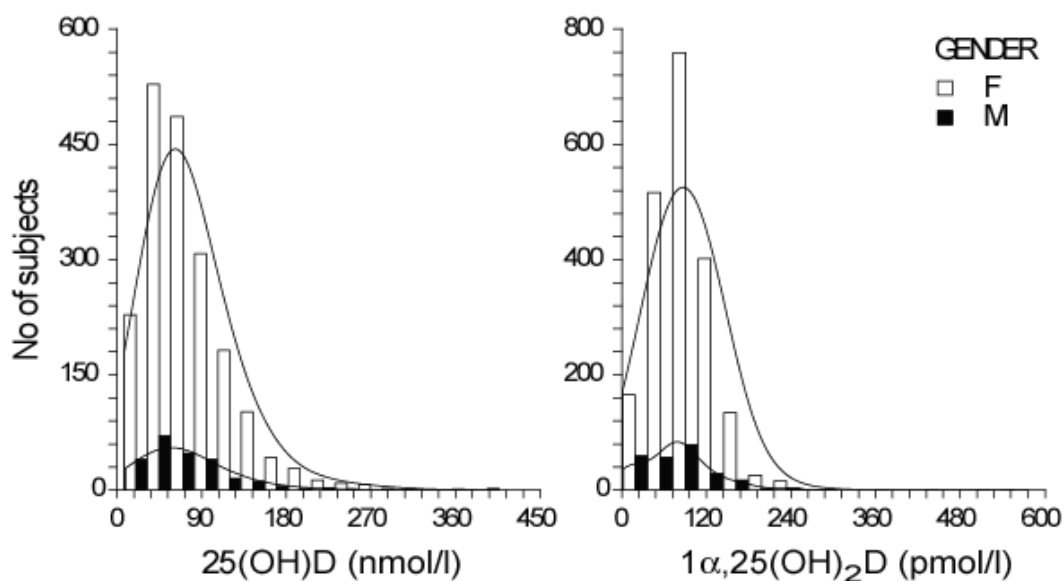


Fig 1 The distribution of serum 25(OH)D and 1α,25(OH)<sub>2</sub>D (F= female, M= male).

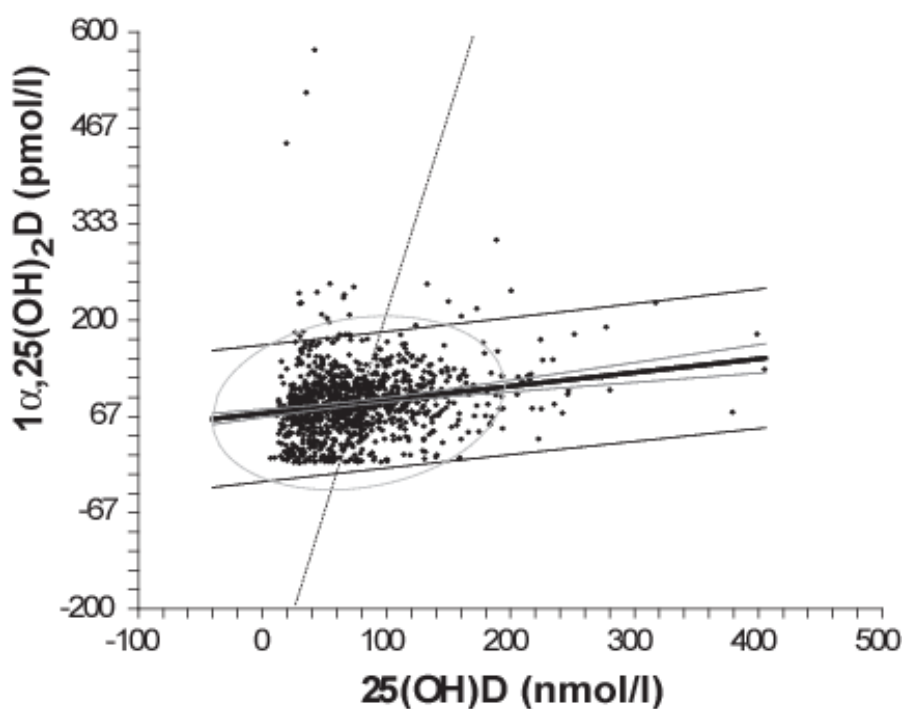


Fig 2 The relationship ( $Y=71.1+0.19*X$ ) between parallel determination of serum 25(OH)D and 1α,25(OH)<sub>2</sub>D (n = 1065, 928 women, 137 men) with showed Y on X line, X on Y line (dotted), probability ellipse, 95% confidence limits and prediction limits.

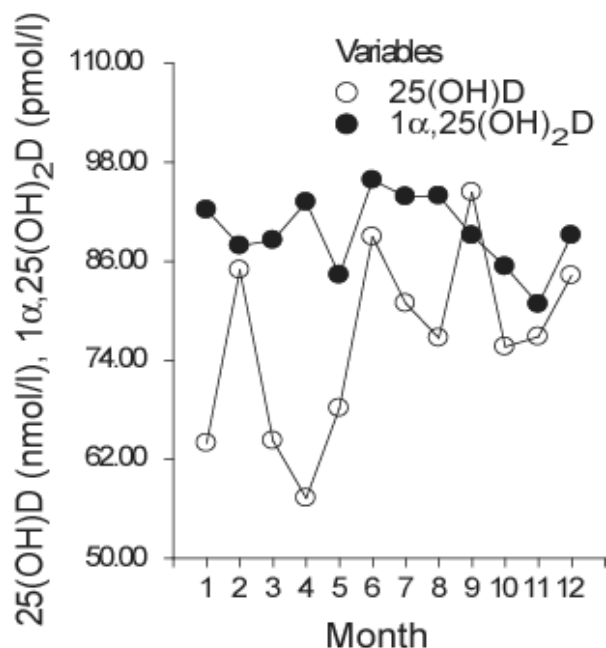


Fig 3 Seasonal variations of serum 25(OH)D and 1α,25(OH)<sub>2</sub>D.

diagnosis, the lowest mean concentrations of 25(OH)D (n=188, 58.9 +/- 38.6 nmol/l) or 1α,25(OH)<sub>2</sub>D (n=224, 53.8 +/- 53.5 pmol/l) were found in patients with hyperparathyroidism, and the highest mean concentrations of 25(OH)D (n=242, 89.2 +/- 55.2 nmol/l) or 1α,25(OH)<sub>2</sub>D (n=234, 100.0 +/- 41.4 pmol/l) in patients treated with cholecalciferol (group of patients with osteoporosis and vitamin D deficiency).

**HPLC data.** The extraction effectiveness of 0.5 ml of serum containing <sup>3</sup>H-25(OH)D<sub>3</sub> with 2 ml acetonitrile was 91 % and reduction of the acetonitrile volume to 1 ml resulted in a decrease in extraction effectiveness to 88 %.

RP HPLC separation of vitamin D metabolites and vitamins A, E, K<sub>1</sub> is shown in Fig. 4. In view of the subsequent radioimmunoanalytical detection of 25(OH)D, the chromatographic parameters of the system were adjusted, so that the retention times of 25(OH)D<sub>2</sub> and 25(OH)D<sub>3</sub> were the same. Fig. 4 shows that within the RP HPLC system used the analytical sensitivity of the UV detector working at 254 nm is insufficient for the measurement of real serum concentrations not only of 1α,25(OH)<sub>2</sub>D, but also of 25(OH)D.

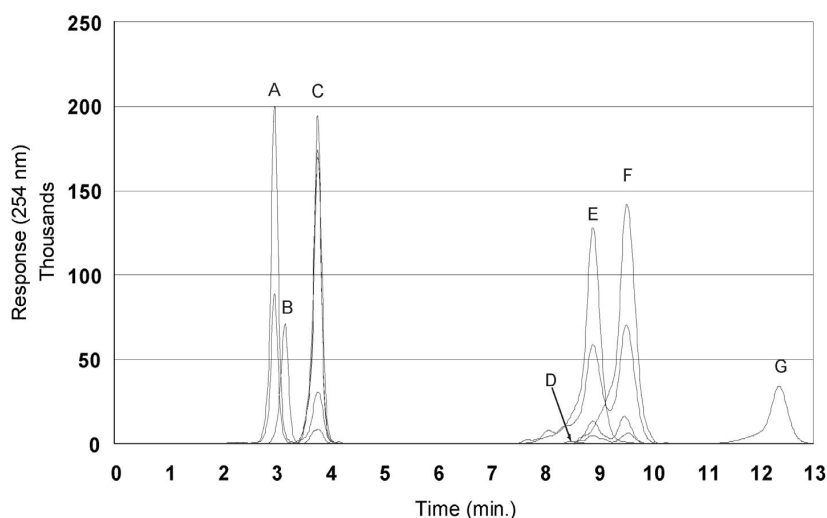
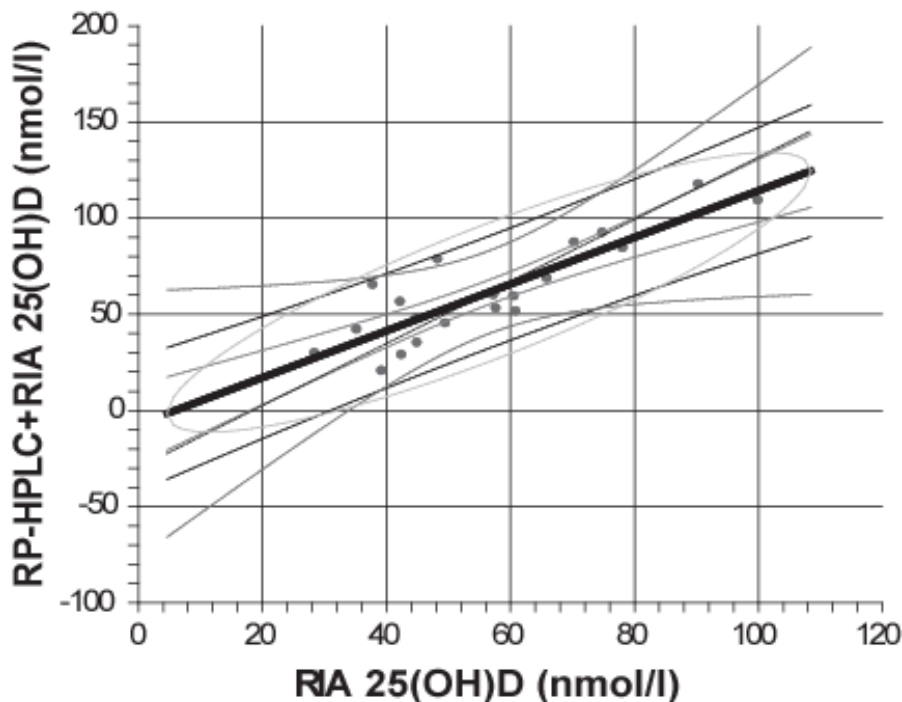


Fig 4 RP HPLC separation of standard solutions of 1α,25(OH)<sub>2</sub>D<sub>3</sub> (10 a 3 μg/100 μl, t<sub>r</sub> = 3,0 min, peak A), vitamin A (5 μg/100 μl, t<sub>r</sub> = 3,2 min, peak B), 25(OH)D<sub>2</sub> and 25(OH)D<sub>3</sub> (10, 2, 0.4 μg/100 μl, t<sub>r</sub> = 3,8 min, t<sub>r</sub> = 3,9 min, peak C), vitamin E (5 μg/100 μl, t<sub>r</sub> = 8,7 min, peak D), ergocalciferol (10, 5, 1, 0.2 μg/100 μl, t<sub>r</sub> = 8,9 min, peak E), cholecalciferol (10, 5, 1, 0.2 μg/100 μl, t<sub>r</sub> = 9,6 min, peak F) and vitamin K<sub>1</sub> (5 μg/100 μl, t<sub>r</sub> = 12,4 min, peak G). The separation was performed isocratically using reverse stationary phase (C18, 250 x 4 mm i.d.) and mobile phase which consists of acetonitrile - methanol 95:5 (flow rate 1,5 ml/min). Standards were detected by UV detector working at 254 nm.



**Fig 5** Scatter plot representing parallel determination ( $n=20$ ) of serum 25(OH)D by commercially available RIA kit from IDS (X axis) and by RP HPLC isolation of 25(OH)D from serum with subsequent RIA quantitation of separated 25(OH)D using the same RIA kit (Y axis). There is showed probability ellipse, 95 % confidence limits and prediction limits. The results are described by the equation  $Y=1.21*X-7.1$ .

Detection of vitamin D metabolites at 290 nm wavelength was approximately 1/3 less effective as compared to that using 254 nm.

In 20 healthy volunteers, parallel determination of serum 25(OH)D was conducted by IDS RIA kit according company instruction, or in acetonitrile extract of serum, where 25(OH)D was isolated by RP HPLC and determined using the same IDS RIA kit. The results are shown in Fig. 5, and are described by the equation  $(HPLC+RIA)=1.2144*RIA-7.1000$  with Pearson's correlation coefficient of 0.867, Spearman's correlation coefficient of 0.7820 and the coefficient of determination  $R^2 = 0.7529$ . The difference between both types of assays was highly significant ( $p<0.0000$ ). The results obtained following HPLC separation ( $61.6 \pm 26.6$  nmol/l) were by 8 % higher than those obtained using the RIA kit only ( $56.6 \pm 19.0$  nmol/l).

### Discussion

Quantification of vitamin D and its metabolites in body fluids belongs to very difficult analytical proce-

dures of clinical chemistry. The reasons for this include the low concentrations of circulating metabolites, their chemical similarities, their physico-chemical instability (ultraviolet light, heat) and the presence of a large number of interfering substances within the biological matrix (especially of neutral lipids). This means that before the analysis itself it is necessary to use separation techniques, such as extraction with liquid solvents, extraction on solid phases, various chromatographic isolation procedures based on normal or reverse phase or on affinity chromatography etc. (Luque-Garcia and Luque de Castro 2001).

In our laboratory we use the extraction of (25(OH)D) and immuno-extraction (FRASER et al. 1997 ) of ( $1\alpha,25(OH)_2D$ ) by commercial RIA kits from IDS (England) for determining 25(OH)D and  $1\alpha,25(OH)_2D$ . In these kits, however, company data on the specific antibody manufactured by IDS does not show the same response to  $25(OH)D_2$  (75 %) as that to  $25(OH)D_3$  (100 %). Recovery of  $25(OH)D_2$  from samples of serum is only 21-29 % (HOLLIS 2000), and there, according to company data, also exists more than 100 % cross-

reactivity to  $24,25(\text{OH})_2\text{D}_3$  in the case of the antibody specific against  $25(\text{OH})\text{D}$ . However, the plasma concentration of  $24,25(\text{OH})_2\text{D}_3$  measured by RP HPLC with UV detection can be higher than 40 nmol/l (BOYER et al. 1999) and thus the circulating  $24,25(\text{OH})_2\text{D}_3$  can substantially contribute to the immunoanalytically measured serum concentration of  $25(\text{OH})\text{D}$ . Similarly, immuno-extraction RIA determination of circulating  $1\alpha,25(\text{OH})_2\text{D}$  from IDS does not show any equivalent response to  $1\alpha,25(\text{OH})_2\text{D}_2$  (91 %) compared to  $1\alpha,25(\text{OH})_2\text{D}_3$  (100 %) (company data). Results of RIA assays of  $25(\text{OH})\text{D}$  ( $n = 2175$ ) shown in Table 1 show that, in accordance with the normal reference range, only 4 % of individuals were inadequately supplied with  $25(\text{OH})\text{D}$ , 79 % were supplied adequately and almost 17 % were supplied more than adequately. Results of RIA assays of  $1\alpha,25(\text{OH})_2\text{D}$  ( $n = 1580$ ) show inadequate concentrations of this main metabolite of vitamin D in 15 % of the individuals tested, 59 % of these individuals were within the reference range and 26 % had values exceeding the upper limit of the reference range. If we were to take into consideration the published comparison of parallel serum determination of serum  $25(\text{OH})\text{D}_3$  ( $n=140$ ) using the HPLC-ESI-MS (HPLC-electro-spray ionisation mass spectrometry) system or the RIA kit from IDS, there is shown that the RIA kit yielded in average by 25 % lower concentration of  $25(\text{OH})\text{D}_3$  (47.8 nmol/l) than the determination by HPLC-ESI-MS (56.8 nmol/l) (QUESADA et al. 2004). From this view the proportion of individuals with more than adequate level of  $25(\text{OH})\text{D}$  would be even higher.

This is one of the reasons why we attempted to compare, in a small group of healthy volunteers ( $n=20$ ), the results obtained by the RIA kit from IDS with those obtained following the extraction, subsequent RP HPLC separation of  $25(\text{OH})\text{D}$  and detection of  $25(\text{OH})\text{D}$  using the same RIA kit from IDS. The correlation is shown in Fig. 5. Similarly as shown by QUESADA et al. (2004), though not to the same extent, the results obtained following HPLC separation were by 8 % higher than those obtained only by the RIA kit. As shown in Fig. 4, by adjusting appropriately the chromatographic conditions, it is possible to separate not only the various hydroxylated metabolites of vitamin  $\text{D}_2$  and  $\text{D}_3$  but also other lipophilic vitamins. Their quantitative determination requires the detection with a higher degree of sensitivity than available by the system of UV detectors that we used. The HPLC/MS combination can be optimal for this type of analysis.

The concentration of both vitamin D metabolites was higher in women than in men (see Table 1), but the statistically significant differences were found only between the mean and median levels of  $1\alpha,25(\text{OH})_2\text{D}$  ( $p<0.0000$ ). Unfortunately, the non-equivalence exists in both groups, whereby the number of men was approximately 8 times lower than that of women. The correlation between parallel determination of  $25(\text{OH})\text{D}$  and  $1\alpha,25(\text{OH})_2\text{D}$  was highly significant ( $p<0.0000$ ), but Figure 2 shows that the dispersion of values was relatively high. Fig. 3, which depicts the changes in  $25(\text{OH})\text{D}$  and  $1\alpha,25(\text{OH})_2\text{D}$  concentrations depending on the individual months of blood sampling, shows that minimum values for both vitamin D metabolites were reached in the late winter season. This apparently corresponds to the extent of sunlight exposure. However, the variance of  $25(\text{OH})\text{D}$  is much higher than the variance of  $1\alpha,25(\text{OH})_2\text{D}$ , and there is a notable peak of  $25(\text{OH})\text{D}$  in the February. This fact cannot be explained by the uneven distribution of proband numbers. The lowest mean concentrations of  $25(\text{OH})\text{D}$  or  $1\alpha,25(\text{OH})_2\text{D}$  were found in patients with hyperparathyroidism.

It is known that a negative relationship exists between serum  $25(\text{OH})\text{D}$  and serum PTH (LIPS 2006), but the situation concerning  $1\alpha,25(\text{OH})_2\text{D}$  is more complicated, because the production of  $1\alpha,25(\text{OH})_2\text{D}$  itself is stimulated by PTH. However, primary hyperparathyroidism and vitamin D insufficiency are common conditions that can occur in combination (MOOSGAARD et al. 2005). Also a secondary hyperparathyroidism frequently observed in patients with chronic kidney disease can result in impaired  $1\alpha$ -hydroxylation of  $25(\text{OH})\text{D}$  to  $1\alpha,25(\text{OH})_2\text{D}$  with concomitantly decreased circulating  $1\alpha,25(\text{OH})_2\text{D}$  levels (COZZOLINO et al. 2006). In accord with such assumption is that the highest mean concentrations of  $25(\text{OH})\text{D}$  or  $1\alpha,25(\text{OH})_2\text{D}$  were found in patients treated with cholecalciferol.

The interpretation of existing and future studies dealing with vitamin D stores must be based on precise and correct analytical procedures. For this reason, clinical assays for determining  $25(\text{OH})\text{D}$  and  $1\alpha,25(\text{OH})_2\text{D}$  should be validated by HPLC-MS, or eventually MS/MS procedures (BOUILLON et al. 2005a). It may be also necessary to reevaluate the recommended intake of vitamin D. HEANEY (2005) stated that the complex realization of vitamin D biological functions is ensured only at concentrations of circulating  $25(\text{OH})\text{D}$  that are approximately equal to 80 nmol/l. This corresponds to an intake of more than 2200 IU of vitamin D daily.



Though this value exceeds the upper limit of the tolerated vitamin D intake (2000 IU), it is associated with optimal absorption of calcium from the intestines and a minimization of increased parathyroid gland activity which is characteristic of the aging population. Osteoporosis induced fractures are reduced and sufficient local production of 1,25(OH)<sub>2</sub>D is ensured (HEANEY 2005). Table 1 shows that this optimal state as expressed by the average level (n = 2175) of circulating 25(OH)D equal to 76.4 nmol/l was achieved in the group studied.

The question is, what is the influence of other vitamin D metabolites and conjugates on the concentration of 25(OH)D measured using commercially available kits. A few is known about this phenomena.

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